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An α2,3-Sialyltransferase from Photobacterium sp. JT-ISH-224 Transfers ^N-Acetylneuraminic Acid to Both the O-2 and O-3' Hydroxyl Groups of Lactose

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An *α*2,3-Sialyltransferase from *Photobacterium* sp. JT-ISH-224 Transfers *N*-Acetylneuraminic Acid to Both the O-2 and O-3 $^{\prime}$ Hydroxyl Groups of Lactose

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We found that *α*2,3-sialyltransferase from *Photobacterium* sp. JT-ISH-224 produced a regio-mistaken sialyl-transferred by-product. Spectroscopic analysis of the purified by-product indicated that it contained two *N*-acetylneuraminic acids: one attached to the O-3' hydroxyl group of lactose, and the other attached to the O-2 hydroxyl group of lactose. The relative configuration between the C-1 and C-3 of the *α*-glucopyranose residue is superimposable with that between C-4 and C-2 of galactopyranoside. Therefore, formation of this by-product, designated 2,3'-disialyllactose, was simply rationalized as a regio-mistaken reaction of bacterial *α*2,3-sialyltransferase. This finding indicates that this bacterial *α*2,3-sialyltransferase has a possibility to synthesize several unusual sialosides.

Keywords 2,3'-Disialyllactose; Regio-mistaken reaction; Sialyltransferase

Abbreviations

CMP-Neu5Ac, cytidine-5- -monophospho-*N*-acetylneuraminic acid; COSY, correlation spectroscopy; Gal, galactose; Glc, glucose; GlcNAc, *N*-acetylgluco-

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samine; HMBC, hetero-nuclear multiple quantum coherence spectrum; HO-HAHA, homo-nuclear hartmann-harn spectrum; HPLC, high-performance liquid chromatography; HSQC, hetero-nuclear single quantum coherence spectrum; Neu5Ac, *N*-acetylneuraminic acid; NMR, nuclear magnetic resonance; TLC, thin layer chromatography; UDP, uridine diphosphate.

INTRODUCTION

Sialic acids are an important component of carbohydrate chains and are usually linked to the terminal positions of the carbohydrate moiety of glycoconjugates such as glycoproteins and glycolipids.^[1,2] Many reports have indicated that sialylglycoconjugates play significant roles in many biological processes, including immunological responses, viral infections, cell–cell recognition, and inflammation.[3,4]

Sialylglycoconjugates are formed *in vivo* by specific sialyltransferases. All sialyltransferases use cytidine-5- -monophospho-*N*-acetylneuraminic acid (CMP-Neu5Ac) as the common donor substrate. Sialyltransferases have been cloned from various sources, including mammalian organs, bacteria, and viruses.[5,6] It has been reported that sialyltransferases of bacterial origin have acceptor substrate specificity that is much broader than that of their mammalian counterparts.^[6,7] Furthermore, several methods for the mass production of bacterial sialyltransferases have been reported.^[8,9] From these points, it is thought that bacterial enzymes will be a valuable tool for the preparation of many kinds of sialosides.^[7,10] As described earlier, the sialylated carbohydrate chains of glycoconjugates play significant roles in many biological processes. Thus, preparation of sialosides is indispensable to investigate their function in detail.

The crystal structures of six sialyltransferases have been reported so far ^[11–16] These sialyltransferases are all of bacterial origin. The first structure described was that of the bifunctional enzyme *α*2,3-/*α*2,8-sialyltransferase (CstII) from *Campylobacter jejuni* OH4384 in complex with a substrate analog.^[11] The second was that of the multifunctional enzyme α 2,3sialyltransferase (Δ 24PmST1) from *Pasteurella multocida* strain P-1059 in the presence and in the absence of CMP.^[12] This enzyme shows α 2,3sialyltransferase activity, *α*2,6-sialyltransferase activity, sialidase activity, and trans-sialidase activity.^[17] The third structure described was that of monofunctional *α*2,3-sialyltransferase (CstI) from *C. jejuni* in apo form and substrate analog-bound form.^[13] The fourth was that of monofunctional α 2,6sialyltransferase (\triangle 16pspST6) from a *Photobacterium* sp. in complex with CMP as a donor product and lactose as the acceptor substrate.^[14] The fifth was that of *α*2,3-sialyltransferase from *P. multocida* in complex with CMP-3FNeuAc and lactose.^[15] The sixth was that of α 2,3-sialyltransferase $(\triangle Npp23ST)$ from *Photobacterium phosphoreum* in complex with CMP.^[16]

Those crystal studies also discussed the substrate binding sites and the catalytic mechanism of the enzymes.

Previously, we reported the cloning *α*2,3-sialyltransferase from *Photobacterium* sp. JT-ISH-224 and the characterization of an N-terminus truncated enzyme.^[10] Hereafter we designate this truncated enzyme as $\Delta23$ pspST3. While using $\Delta 23$ pspST3 to produce 3[']-sialyllactose, we found a by-product in the enzymatic reaction mixture. In this by-product, sialic acid was incorporated at an unusual position through regio-mistaken transfer. In addition to the knowledge provided by x-ray analysis data, this regio-mistaken transfer provides insight into the mode of binding with acceptors in bacterial sialyltransferase. We report herein on the structure of the by-product. The results of this study open the possibility of using this marine bacterial α 2,3-sialyltransferase to produce sialosides of unusual structure.

RESULTS AND DISCUSSION

During the course of purifying 3'-sialyllactose, we found a by-product in the enzymatic reaction mixture. Analysis by HPLC revealed that retention time of the by-product was different from that of 3'-sialyllactose (by-product, 29.5) minutes; 3'-sialyllactose, 21.8 minutes). Furthermore, analysis by thin layer chromatography (TLC) suggested that the by-product was a disialylated saccharide. Each compound was then purified further by size-exclusion column chromatography; 140.3 mg of 3'-sialyllactose and 9.3 mg of by-product were isolated. The yield of 3'-sialyllactose and the by-product based on the donor substrate (353 μ mol) was 62.1% and 5.6%, respectively. After this purification, we analyzed the by-product by NMR and mass spectrometry to determine its structure. The mass spectrum of the by-product showed an $(M-H)^+$ ion peak at 923.5. Because the expected mass of lactose having two *N*-acetylneuraminic acids in its structure is 924.3, this result strongly indicated that the by-product possessed two *N*-acetylneuraminic acids. Next, we analyzed the by-product by ¹H-NMR spectroscopy. Interestingly, in the ¹H-NMR spectrum of this byproduct (Fig 1) we observed no glucose residue *β*-anomer signals (for example, *δ* 3.2 ppm for Glc *β* H-2, *δ* 4.6 ppm for Glc *β* H-1, and so on, as were observed for 3',6'-disialyllactose; Kajihara et al.^[18]); furthermore, the H-2 chemical shift of Glc was shifted downfield by 0.3 ppm compared to that of 3'-sialyllactose, indicating that this by-product consisted only of the *α*-form. The ¹H and ¹³C chemical shifts obtained from HSQC, HMBC, HOHAHA, and COSY spectra are summarized in Table 1. The sialylated position was identified from the differences in the ¹³C chemical shift values observed between the by-product and 3',6'-sialyllactose.^[18] The C-2 chemical shift of the glucose of the by-product was downfield, and we observed a correlation between the C-2 of Neu5Ac and the H-2 of Glc, and a correlation between the C-2 of Neu5Ac and the H-3 of

Figure 1: 1 H-NMR spectrum of the purified by-product. The analysis conditions are described in Materials and Methods. The chemical shift is expressed relative to internal HOD (4.75 ppm).

Gal (Fig 2). From these results, it was clear that a Neu5Ac was attached to the O-3' hydroxyl group and that another Neu5Ac was attached to the O-2 hydroxyl group of lactose. The structure of the by-product, designated 2,3- disialyllactose, is shown in Figure 3.

Although ours is the first report regarding a regio-mistaken reaction by sialyltransferase, a similar type of mistaken glycosyltransferase reaction was reported by Nishida et al. for galactosyltransferase.^[19] Nishida et al. reported that although galactosyltransferase prepared from bovine milk usually transfers the galactose residue of UDP-galactose to the OH-4 positions of glucose and *N*-acetylglucosamine (GlcNAc), the galactosyltransferase also catalyzed the transfer of *β*Gal from UDP-galactose to the anomeric position of *N*-acetylkanosamine (Glc3NAc). In this case, the stereochemistry of the *β*anomer of Glc3NAc from C-1 to C-4 is identical to that of C-4 to C-1 of GlcNAc. Thus, galactosyltransferase recognizes Glc3NAc as an acceptor substrate and transfers galactose to the *β*-anomeric position of the Glc3NAc, which corresponds to C-4 of GlcNAc.^[19] With lactose, which was the acceptor substrate used in our current study, the stereochemistry of *α*-glucose from C-1 to C-3 is superimposable with that of C-4 to C-2 of galactose. This fact strongly indicates

	Neu ₅ Ac			Gal			Glc		
	ŀΗ	J(Hz)	${}^{13}C$	¹ Η	J(Hz)	${}^{13}C$	¹ Η	J(Hz)	${}^{13}C$
$\begin{array}{c} \hline \end{array}$ $\overline{2}$			176.3 102.5	4.49 3.52	d, 7.85	104.9 71.8	5.19 (α) 3.95 (α)	d, 3.53 dd, 9.82, 3.53	93.1 75.9
$\mathbf{2}$ 3ax	1.75	dd, 12.63	102.2 41.7	4.06	dd, 9.90, 3.06	77.9	$3.79(\alpha)$		72.1
3ax 3ea	2.67	dd, 12.63, 4.65	42.1 41.7						
3eq	2.70	dd, 12.49, 4.54	42.1						
4	3.62		71.0 70.8	3.90		69.9	$3.69(\alpha)$		80.1
$\frac{4}{5}$ 6 $\begin{array}{c} 6 \\ 7 \end{array}$ $\overline{7}$	3.79 3.56 3.53		54.1 75.3 75.2 70.6 70.5	3.65 3.72 3.65		77.6 63.5	$3.89(\alpha)$ 3.82 (α)		72.4 62.4
$\begin{matrix}8\\8\end{matrix}$ 9	3.85 3.82 3.58		74.2 74.2 65.0						

Table 1: Assignments of H and H^3C chemical shift and coupling constant (J) of the purified by-product

The chemical shifts of 1 H and 13 C NMR are expressed relative to internal HOD = 4.75 ppm (298) K) and external 1,4-dioxane $= 69.1$ ppm (298 K), respectively.

that the marine bacterial α 2,3-sialyltransferase recognized the C-1 to C-3 part of the *α*-glucose as an acceptor substrate site and transferred Neu5Ac from CMP-Neu5Ac to the O-2 hydroxyl group of *α*-glucose. In the case of *β*-glucose, the configuration of the hydroxyl groups from C-1 to C-3 is not superimposable with that of C-4 to C-2 of galactose.

Although the reason why the by-product is fixed to *α*-anomer is still unclear, we speculate that after transferring Neu5Ac to the O-2 hydroxyl group of *α*-glucose, the hydrogen bond between the carboxyl group of Neu5Ac and the O-1 hydroxyl group of *α*-glucose might immediately form and stabilize the molecule.

*-*23pspST3 used in this study was a truncated recombinant enzyme cloned from *Photobacterium* sp. JT-ISH-224. This enzyme was classified according to amino acid sequence similarities into Glycosyltransferase Family 80 (GT80) in the CAZy database.^[20] Of the six sialyltransferases whose crystal structures have been determined so far, four are classified into GT80. These four are the multifunctional *α*2,3-sialyltransferase ($Δ24PmST1$) from *P. multocida*

Figure 2: HMBC spectrum of the purified by-product. The analysis conditions are described in Materials and Methods.

strain P-1059,^[12] the α 2,6-sialyltransferase (Δ 16pspST6) from a *Photobacterium* sp.,^[14] the *α*2,3-sialyltransferase from *P. multocida*,^[15] and the *α*2,3sialyltransferase ($\triangle Npp23ST3$) from *P. phosphoreum*.^[16] Although the crystal structure of the *α*2,3-sialyltransferase cloned from *Photobacterium* sp. JT-ISH-224 has not yet been solved, the ternary structure of the enzyme seems to closely resemble that of $\Delta 24$ PmST1 owing to their close similarities in terms of amino acid sequence (29% identity). In the crystal structure of $\Delta 24\mathrm{PmST1}$ in complex with CMP and lactose, the lactose was recognized by five amino acid residues, $Trp270$, Asp141, His112, Asn85, and Met144.^[21] Each of those residues contacted only the galactose part of the lactose. However,

Figure 3: Structure of the by-product.

in *A23*pspST3, only one of those amino acid residues, Asp141, is conserved (as Asp125). Pair-wise alignment $\Delta 24$ PmST1 and $\Delta 23$ pspST3 show that the amino acid residues corresponding to the Trp270, His112, Asn85, and Met144 of $\Delta 24\mathrm{PmST1}$ are all substituted or deleted in $\Delta 23\mathrm{pspST3}$. We are presently conducting a number of studies, including a mutational study and a structural study of the enzyme, to clarify the mechanisms of acceptor substrate recognition and binding of this enzyme.

The results obtained in this study indicate the possibility for production of several sialosides having unique structures. Although the acceptor substrate binding mechanism of this enzyme is still unclear, compounds having a stereochemical structure similar to that of *α*-glucose from the C-1 to C-3 part are likely to be candidate acceptor substrates for this enzyme. We are currently investigating the possibility of synthesizing unique sialosides by means of this bacterial sialyltransferase.

MATERIALS AND METHODS

Materials

CMP-Neu5Ac was purchased from Japan Food and Liquor Alliance (Kyoto, Japan); lactose was from Wako Pure Chemical Industries (Osaka, Japan); AG 1-X2 was from Bio-Rad Laboratories (Hercules, CA, USA); Sephadex G-15 was from GE Healthcare Bioscience (Tokyo, Japan); TLC plates were from Merck (Darmstadt, Germany). All other reagents were of commercially available analytical grade.

Preparation of recombinant *α*2,3-sialyltransferase from *Photobacterium* sp. JT-ISH-224

The recombinant *α*2,3-sialyltransferase used in this study was prepared using the methods reported by Tsukamoto et al.^[22] One unit (U) of sialyltransferase was defined as the amount of enzyme required to transfer 1μ mol of Neu5Ac from CMP-Neu5Ac to lactose in 1 min.

Enzymatic Synthesis of Sialyloligosaccharides

The enzymatic synthesis was performed using purified recombinant *α*2,3 sialyltransferase. The reaction mixture was composed of 100 mg of lactose (280 *µ*mol), 250 mg of CMP-Neu5Ac (353 *µ*mol), and 5.05 U of purified *α*2,3 sialyltransferase from *Photobacterium* sp. JT-ISH-224 in 0.5 mL of 100 mM Bis-Tris buffer (pH 6.0), and was incubated at 30° C for 2 h.

Purification of Sialyloligosaccharides

The enzymatic reaction products, including the main product and the byproduct, were purified according to the method of Sabesan and Paulson^[23] with a slight modification. After the reaction, the reaction mixture was diluted with 10 mL of deionized water and introduced into an Econo column (1.0 cm \times 10 cm, Bio-Rad Laboratories) containing AG 1-X2 ion-exchange resin (phosphate form, 200–400 mesh). The column was washed with deionized water, and then the product was eluted twice with 10 mL each of 5, 10, 50, 100, 500, or 1000 mM potassium phosphate buffer (pH 6.8). An aliquot of each eluted fraction was analyzed by TLC, as described later, and the fractions containing glycosidic Neu5Ac were evaporated to dry residues. Each of the dry residues was dissolved in 2.5 mL of deionized water and then loaded onto a Sephadex G-15 column (1.6 cm \times 70 cm). The product was eluted with deionized water. The fractions containing glycosidic Neu5Ac were pooled and evaporated to dryness.

Analysis of Sialyloligosaccharides

Sialyloligosaccharides formed by the enzymatic reaction were analyzed by TLC or HPLC (Shimadzu, Kyoto, Japan) with a TSK-gel amide-80 column (4.6 $mm \times 25$ cm; Tosoh, Tokyo, Japan). In the case of TLC analysis, a small amount of the enzymatic reaction mixture was applied to a precoated silica gel plate (60 F254, Merck) and developed with a solvent consisting of 2-propanol/acetic acid/water (3:2:1). For visualization of the organic compounds, the plate was dipped into a solution of 5% sulfuric acid in ethanol and then heated. HPLC analyses were performed according to the method described by Endo et al.^[24] The purified products were analyzed by ESI-MASS spectroscopy as in a previous report^[10] and by NMR spectroscopy using a Bruker DMX-500 spectrometer (Bruker Billerica, MA, USA) at 298 K in D_2O . The ${}^{1}H-{}^{13}C$ single-bond and multibond correlation spectra were measured by use of the HSQC, HMBC, HO-HAHA, and COSY methods with the standard Bruker programs. The molecular mass of the by-product was determined by mass spectrometry.

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